

# Compilation of Historical Local Lymph Node Data for Evaluation of Skin Sensitization Alternative Methods

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**Background:** Within the toxicology community, considerable effort is directed toward the development of alternative methods for skin sensitization testing. The availability of high-quality, relevant, and reliable *in vivo* data regarding skin sensitization is essential for the effective evaluation of alternative methodologies. Ideally, data derived from humans would be the most appropriate source because the test methods are attempting to predict a toxicologic effect in humans. Unfortunately, insufficient human data of the necessary quality are available, so it is necessary to rely on the best available animal data. In recent years, the local lymph node assay (LLNA) has emerged as a practical option for assessing the skin sensitization potential of chemicals. In addition to accurately identifying skin sensitizers, the LLNA can also provide a reliable measure of relative sensitization potency, information that is pivotal to the successful management of human health risks.

**Objective:** To provide a database of robust *in vivo* data to calibrate, evaluate, and eventually validate new approaches for skin sensitization testing.

**Methods:** LLNA data derived from previously conducted studies were compiled from the published literature and unpublished sources.

**Results:** We provide a database that comprises LLNA data on 211 individual chemicals. This extensive chemical data set encompasses both the chemical and biologic diversity of known chemical allergens. To cover the range of relative allergenic potencies, the data set includes data on 13 extreme, 21 strong, 69 moderate, and 66 weak contact allergens, classified according to each allergen's mathematically estimated concentration of chemical required to induce a threefold stimulation index. In addition, there are also 42 chemicals that are considered to be nonsensitizers. In terms of chemical diversity, the database contains data pertaining to the chemical classes represented by aldehydes, ketones, aromatic amines, quinones, and acrylates, as well as compounds that have different reactivity mechanisms. In addition to two-dimensional chemical structures, the physicochemical parameters included are log K<sub>p</sub>, log K<sub>O/W</sub>, and molecular weight.

**Conclusions:** The list of chemicals contained in the data set represents both the chemical and biologic diversity that is known to exist for chemical allergens and non-allergens. It is anticipated that this database will help accelerate the development, evaluation, and eventual validation of new approaches to skin sensitization assessment.

THE EVALUATION OF THE POTENTIAL to induce allergic contact dermatitis (ACD) is an important component of the safety assessment process for a new chemical that may be encountered through the skin.<sup>1</sup> For many years, the guinea pig has been the species of choice

for the identification of skin-sensitizing hazards. More recently, however, the local lymph node assay (LLNA) has been accepted as a valid alternative approach; its method is based on the characterization of induced proliferative responses in the draining lymph nodes after topical exposure of mice to chemicals.<sup>2-6</sup> The LLNA has been adopted by the Organization for Economic Cooperation and Development (OECD) as a stand-alone test for skin sensitization (OECD Test Guideline 429).<sup>7</sup> This adoption resulted from thorough and independent validations of the LLNA in both the United States<sup>8</sup> and Europe.<sup>9</sup> The LLNA offers clear scientific advantages as well as important animal welfare benefits (in terms of both reduction and refinement) when compared with traditional guinea pig tests. In addition to accurately identifying skin sensitization hazards, the LLNA can provide a reliable measure of relative allergenic potency,

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information that is critical to the risk assessment process. This is achieved by consideration of the vigor with which chemical allergens provoke proliferative responses in draining lymph node cells. The relative potency of a chemical sensitizer is measured by deriving what is known as the EC<sub>3</sub> value, which indicates the test chemical concentration necessary to produce a threefold stimulation of proliferation in draining lymph nodes as compared with concurrent vehicle controls.<sup>10</sup> Thus, the intrinsic sensitizing potency of a chemical is defined as a function of the concentration required to elicit a threshold positive response, a stimulation index (SI) equal to 3. For this reason, the lower the EC<sub>3</sub> value, the greater the relative skin sensitizing potency of the chemical. This approach has already been applied with some success, and the LLNA is the preferred method for estimating the potency of a contact allergen.<sup>11,12</sup>

Meeting the ultimate challenge of developing non-animal predictive test methods to replace an *in vivo* skin sensitization test such as the LLNA requires an understanding of the biologic mechanisms involved in the development of ACD.<sup>13,14</sup> Specifically, the *in vitro* approach should be designed in such a way that the potential of a chemical to penetrate the skin, react with protein/peptide (with or without the need for biotransformation), and initiate an antigen-specific immune response are incorporated into the developed tests. Approaches could include *in vitro* methods, based most commonly on chemically induced responses in cell culture systems (eg, single-cell cultures or more complex models such as skin explant and skin-equivalent models), or *in silico* methods (eg, computer-based expert or quantitative structure activity systems).

Currently, *in vitro* systems for the identification of skin sensitizers are still in their infancy. One critical requirement for the development of alternative test methods will be the use of robust *in vivo* data to calibrate, evaluate, and eventually validate new approaches. In this article, we describe the compilation of an extensive chemical data set that embraces a range of chemicals and skin-sensitizing activity. All materials have been evaluated through LLNA; for some chemicals, it has been demonstrated that the LLNA EC<sub>3</sub> value correlates closely with what is known of the chemical's relative ability to induce sensitization in humans.<sup>15-17</sup> These data provide a unique and valuable list of chemicals for which the sensitivity, selectivity, and overall accuracy of proposed alternative methods for skin sensitization can be judged. This compilation also provides an invaluable source of data with which to explore other issues, such as the relationship between chemical categorization (on the basis of mechanism of action) and potency. It could also help to define the applicability

domain of the LLNA as well as that of existing alternative methodologies and those in development.

## Materials and Methods

### Chemicals

The chemicals identified in this article have been evaluated for skin sensitization potential with the LLNA. For each chemical listed in Table 1, the chemical abstract service (CAS) number and two-dimensional structure are specified. In addition, the molecular weight, the skin penetration coefficient ( $\log K_p$ : centimeters  $\times$  hours  $- 1$ , here expressed in its log scale), and the octanol-water partition coefficient ( $\log K_{O/W}$ ) values are indicated for each chemical. The structures were graphically drawn with *ChemDraw*, version 6.0 (CambridgeSoft, Cambridge, MA). The molecular weight (MW),  $\log K_p$ , and  $\log K_{O/W}$  values were obtained by analysis of the structures with the expert system *DEREK for Windows*, version 7 (Lhasa Limited, Leeds, UK).<sup>18</sup> The  $\log K_p$  was calculated with the equation of Potts and Guy,<sup>19</sup> and the  $\log K_{O/W}$  was calculated with Moriguchi's estimation.<sup>20</sup>

### Local Lymph Node Assay Protocol

The LLNA was conducted as described in the literature (Fig 1).<sup>2-6</sup> Briefly, groups of CBA female mice (7-12 weeks of age) were exposed topically on the dorsum of both ears to 25  $\mu$ L of test material or to an equal volume of the relevant vehicle alone. Treatment was performed daily for 3 consecutive days. Five days after the initiation of exposure, all mice were injected via the tail vein with 250  $\mu$ L of phosphate-buffered saline containing 20  $\mu$ Ci of tritiated thymidine. The mice were sacrificed 5 hours later, and the draining auricular lymph nodes were excised and pooled for each experimental group or each individual animal. The incorporation of tritiated thymidine measured by beta scintillation counting was reported in disintegrations per minute. An SI was calculated for each chemical-treated group as the ratio of the disintegrations per minute in the treated group (or mean disintegrations per minute when individual animals were assessed) to the disintegrations per minute or mean disintegrations per minute of the concurrent vehicle control group. LLNA methodology assesses skin sensitization, not photosensitization. A substance was classified as a skin sensitizer if, at one or more test concentrations, it induced a threefold or greater increase in local lymph node proliferative activity when compared with concurrent vehicle-treated controls (SI,  $\geq 3$ ). The compilation of data from numerous laboratories is reported in

Table 1. Chemical Structures, Physicochemical Parameters, LLNA Data, and Potency Categorization

Chemical Structure	CAS No.	Log K <sub>ow</sub>	MW	Vehicle	% VNTT	LLNA SI	LLNA SI	LLNA SI	LLNA SI	LLNA EC3* (%)	Potency Category	Reference								
Abietic acid	514-10-3	-1.29	4.61	302.46	AOO	5.0	10.0	25.0	—	—	1.5	2	5.2	—	—	15	Weak	Ashby J et al. <sup>21</sup>		
	874-23-7	-2.41	1.66	143.21	Acetone	10.0	20.0	40.0	—	—	0.8	0.7	0.8	—	—	NC	Nonsensitizer	P&G (unpublished)		
	140-67-0	-1.82	2.54	148.21	AOO	10.0	25.0	50.0	100.0	—	1.2	4.7	4.5	8	—	18	Weak	Unilever (unpublished)		
	6358-09-4	-3.69	0.26	188.57	AOO	0.1	0.25	0.5	1.0	2.5	—	1.7	1.4	2.1	1.5	3.4	—	2.2	Moderate	Estrada E et al. <sup>22</sup>
	2871-01-41	-3.84	0.12	197.19	AOO	0.1	0.25	0.5	1.0	2.5	—	0.5	1.2	1.9	3.3	—	2.2	Moderate	Estrada E et al. <sup>22</sup>	

AOO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMF = dimethylformamide; DMSO = dimethyl sulfoxide; EtOH = ethanol; FEMIA = Flavor and Extract Manufacturers' Association; K<sub>ow</sub> = octanol-water partition coefficient (log scale); K<sub>p</sub> = skin penetration coefficient (log scale); MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated; P&G = Procter & Gamble Co.; RIEM = Research Institute for Fragrance Materials.

\*Mathematically estimated concentration of test chemical necessary to induce a threefold stimulation index (SI).

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Table 1. Continued

Chemical Structure	CAS No.	Log K <sub>ow</sub>	MW	Vehicule	LLNA %	IS VNTT	LLNA EC <sub>3</sub> * (%)	Potency Category	Reference
2-Aminophenol	95-55-6	-2.55	1.17	109.13	AOO	0.5	1.0	2.5	—
						—	—	3.5	5.0
3-Aminophenol	591-27-5	-2.55	1.17	109.13	AOO	2.5	5.0	10.0	—
						—	—	2.8	3.5
α-Amyl cinnamic aldehyde	122-40-7	-1.45	3.52	202.30	AOO	1.0	2.5	5.0	10.0
						—	—	25.0	—
2-(4-tert-Amylcyclohexyl)acetaldehyde (QRM 2113)	620159-84-4	-1.59	3.28	196.33	AOO	25.0	50.0	100.0	—
						—	—	2.1	4
trans-Anethol	104-46-1	-1.82	2.54	148.21	AOO	4.5	9.0	22.6	—
						—	—	13.5	24.7
Aniline	62-53-3	-2.18	1.56	93.13	AOO	5.0	10.0	25.0	50.0
						—	—	1.1	0.9
						—	—	2.0	1.9
						—	—	3.3	3.3
						—	—	89	Weak
						—	—	Unilever (unpublished)	Unilever (unpublished)

AOO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMF = dimethylformamide; DMSO = dimethyl sulfoxide; EtOH = ethanol; FEMA = Flavor and Extract Manufacturers' Association; K<sub>ow</sub> = octanol-water partition coefficient (log scale); K<sub>p</sub> = skin penetration coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated; P&G = Procter & Gamble Co.; RIEM = Research Institute for Fragrance Materials.

\*Mathematically estimated concentration of test chemical necessary to induce a threefold stimulation index (SI).

<sup>†</sup>Value estimated.

Table 1. Continued

Chemical Structure	CAS No.	Log K <sub>p</sub>	Log K <sub>ow</sub>	MW	Vehicle	LLNA %	LLNA %	LLNA %	LLNA %	IS VNTT	LLNA EC <sub>3*</sub> (%)	Potency Category	Nonsensitizer	Reference						
Benzaldehyde	100-52-7	-2.09	1.80	106.12	AOO	1.0	2.5	5.0	10.0	25.0	—	2.1	1.7	2.2	1.8	2	—	NC	Pallewicz G et al. <sup>24</sup>	
1,2,4-Benzenetricarboxylic anhydride (trimellitic anhydride)	552-30-7	-3.36	0.75	192.13	AOO	1.0	2.5	5.0	10.0	25.0	—	1.1	2	2.0	3.2	4.6	—	9.2	Moderate	Estrada E et al. <sup>22</sup>
1,2-Benzisothiazolin-3-one (Proxel active)	2634-33-5	-2.64	1.42	151.18	DMF	10.0	30.0	50.0	—	—	—	3.8	4.4	4.9	—	—	—	2.3 <sup>i</sup>	Moderate	Ashby J et al. <sup>21</sup>
Benzo[a]pyrene	50-32-8	-0.44	5.39	252.32	AOO	0.5	1.0	2.5	—	—	—	17.6	19.2	27.0	—	—	—	0.0009 <sup>j</sup>	Extreme	Ashby J et al. <sup>21</sup>
Benzocaine	94-09-7	-2.65	1.52	165.19	AOO	2.5	5.0	10.0	25.0	50.0	—	2.1	1.8	2.7	1.8	1.2	—	NC	Nonsensitizer	Warbrick EV et al. <sup>25</sup>

AOO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMF = dimethylformamide; DMSO = dimethyl sulfoxide; EtOH = ethanol; FFMA = Flavor and Extract Manufacturers' Association; K<sub>ow</sub> = octanol-water partition coefficient (log scale); K<sub>p</sub> = skin penetration coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated; P&G = Procter & Gamble Co.; RIFM = Research Institute for Fragrance Materials.

\*Mathematically estimated concentration of test chemical necessary to induce a threefold stimulation index (SI).

<sup>i</sup>Value estimated.

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Table 1. *Continued*

AOO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMF = dimethyl formamide; DMSO = dimethyl sulfoxide; EtOH = ethanol; FEMA = Flavor and Extract Manufacturers' Association; K<sub>ow</sub> = octanol-water partition coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated;

PP&G = Procter & Gamble Co.; RJFM = Research Institute for Fragrance Materials.

\*Mathematically estimated

Table 1. Continued

Chemical Structure	CAS No.	MW	Log K <sub>ow</sub>	Vehicle	LLNA EC <sup>a</sup> (%)	LLNA EC <sub>3</sub> <sup>a</sup> (%)	Potency Category	Reference											
Bisphenol A-diglycidyl ether	1675-54-3	-1.89	4.09	340.42	AOO	1.0	3.0	10.0	—	—	20	6.0	17.4	—	—	1.5	Moderate	Warbrick EV et al. <sup>29</sup>	
1-Bromobutane	109-65-9	-2.26	1.82	137.02	AOO	5.0	10.0	25.0	—	—	1.1	1.2	1.0	—	—	NC	Nonsensitizer	Ashby J et al. <sup>21</sup>	
1-Bromodocosane	6938-66-5	-0.66	6.25	389.51	AOO	2.5	5.0	10.0	—	—	1.2	1.6	3.7	—	—	8.3	Moderate	Estrada E et al. <sup>22</sup>	
1-Bromododecane	143-15-7	-1.55	3.79	249.24	AOO	5.0	10.0	25.0	—	—	1.1	1.4	4.5	—	—	18	Weak	Baskett DA et al. <sup>30</sup>	
12-Bromododecanoic acid	73367-80-3	-2.28	3.02	279.22	AOO	5.0	10.0	25.0	—	—	1.3	2	3.9	—	—	18	Weak	Unilever (unpublished)	
12-Bromo-1-dodecanol	3344-77-2	-1.92	3.40	265.24	AOO	5.0	10.0	25.0	—	—	—	2.2	4.3	9.8	—	—	6.9	Moderate	Estrada E et al. <sup>22</sup>
1-Bromoetcosane	4276-49-7	-0.84	5.76	361.45	AOO	5.0	10.0	25.0	—	—	2.1	6.2	8.4	—	—	6.1	Moderate	Estrada E et al. <sup>22</sup>	

AO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMF = dimethylformamide; DMSO = dimethyl sulfoxide; EtOH = ethanol; FEMA = Flavor and Extract Manufacturers' Association; K<sub>ow</sub> = octanol-water partition coefficient (log scale); K<sub>p</sub> = skin penetration coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated; P&G = Procter & Gamble Co.; RIFM = Research Institute for Fragrance Materials.

\*Mathematically estimated concentration of test chemical necessary to induce a threefold stimulation index (SI).

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Table 1. Continued

Chemical Structure	CAS No.	Log K <sub>ow</sub>	MW	V <sub>o/w</sub>	LLNA %	LLNA EC <sub>50</sub> * (μg)	Potency Category	Reference												
1-Bromohexadecane	3508-00-7	-1.11	5.02	319.37	AOO	5.0	10.0	25.0	—	—	3.2	6	9.6	—	4.8 <sup>†</sup> Moderate	Baskett DA et al. <sup>10</sup>				
1-Bromoheptadecane	112-82-3	-1.20	4.77	305.34	AOO	1.0	2.5	5.0	10.0	25.0	50.0	1.1	3.3	7.9	11.1	13.5	16.8	2.3	Moderate	Baskett DA et al. <sup>10</sup>
1-Bromoheptadecane	111-25-1	-2.09	2.31	165.07	AOO	1.0	10.0	50.0	—	—	1.7	2.9	18.6	—	—	—	—	10	Weak	Estrada E et al. <sup>22</sup>
3-Bromomethyl-5,5-dimethyl-2-hydro-2(3H)-furanone	154750-20-6	-2.72	1.79	207.07	AOO	3.19	6.37	12.74	—	—	2.7	5.1	7.1	—	—	—	—	3.6	Moderate	Estrada E et al. <sup>22</sup>
1-Bromononane	693-58-3	-1.82	3.05	207.16	AOO	5.0	10.0	25.0	—	—	1.2	1.4	2.8	—	—	—	NC	Nonsensitizer	Baskett DA et al. <sup>10</sup>	
1-Bromoocooctadecane	112-89-0	-1.02	5.26	333.40	AOO	5.0	10.0	25.0	—	—	1.8	2.2	4.5	—	—	—	—	15	Weak	Baskett DA et al. <sup>10</sup>
1-Bromopentadecane	629-72-1	-1.28	4.53	291.32	AOO	5.0	10.0	25.0	—	—	2.9	7.8	19.6	—	—	—	—	5.1	Moderate	Ashby J et al. <sup>21</sup>
1-Bromotetradecane	112-71-0	-1.37	4.28	277.29	AOO	5.0	10.0	25.0	—	—	1.5	3.3	11.3	—	—	—	—	9.2	Moderate	Baskett DA et al. <sup>10</sup>
7-Bromotetradecane	74036-97-8	-1.37	4.28	277.29	AOO	5.0	10.0	25.0	—	—	0.9	1.2	3.6	—	—	—	—	21	Weak	Estrada E et al. <sup>22</sup>

AOO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMF = dimethylformamide; DMSO = dimethyl sulfoxide; EtOH = ethanol; FEMA = Flavor and Extract Manufacturers' Association; K<sub>ow</sub> = octanol/water partition coefficient (log scale); K<sub>p</sub> = skin penetration coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated; P&G = Procter & Gamble Co.; RIFM = Research Institute for Fragrance Materials.

\*Mathematically estimated concentration of test chemical necessary to induce a threefold stimulation index (SI).

<sup>†</sup>Value estimated.

Table I. Continued

Chemical Structure	CAS No.	Log K <sub>ow</sub>	MW	Vehicle	LLNA	EC <sub>3</sub> * (%)	LLNA EC <sub>3</sub> (%)	Potency Category	Reference		
2-Bromotetradecanoic acid	10520-81-7	-2.10	3.51	307.27	AOO	5.0	10.0	25.0	—	IS VNTI	Baskett DA et al. <sup>30</sup>
1-Bromotridecane	765-09-3	-1.46	4.03	263.26	AOO	5.0	10.0	25.0	—	IS VNTI	Baskett DA et al. <sup>30</sup>
1-Bromoundecane	693-67-4	-1.64	3.54	235.21	AOO	5.0	10.0	25.0	—	IS VNTI	Baskett DA et al. <sup>30</sup>
2,3-Butanedione	431-03-8	-2.76	0.68	86.09	AOO	5.0	10.0	25.0	—	% VNTI	Roberts DW et al. <sup>31</sup>
1-Butanol	71-36-3	-2.42	1.06	74.12	dH <sub>2</sub> O	5.0	10.0	20.0	—	% VNTI	Ryan CA et al. <sup>28</sup>
α-Butyl cinnamic aldehyde hydrocinnamal (Lilial)	7492-44-6	-1.54	3.28	188.27	AOO	1.0	2.5	5.0	10.0	IS VNTI	Elahi EN et al. <sup>23</sup>
p-tert-Butyl-a-ethyl hydrocinnamal (Lilial)	80-54-6	-1.46	3.52	204.31	AOO	1.0	2.5	10.0	25.0	IS VNTI	Baskett DA et al. <sup>16</sup>

AOO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMF = dimethylformamide; DMSO = dimethyl sulfoxide; EtOH = ethanol; FFMA = Flavor and Extract Manufacturers' Association; K<sub>ow</sub> = octanol/water partition coefficient (log scale); KP = skin penetration coefficient (log scale); MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated; P&G = Procter & Gamble Co.; RIFM = Research Institute for Fragrance Materials.

\*Mathematically estimated concentration of test chemical necessary to induce a threefold stimulation index (SI).

<sup>†</sup>Value estimated.

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Table 1. *Continued*

Chemical Structure	CAS No.	$\log K_p^{\text{O}_2/W}$	MW	Vehicle	LLNA EC <sub>50</sub> * (%)										Potency Category	Reference			
					LLNA SI	LLNA NTT	LLNA SI	LLNA NTT	LLNA SI	LLNA NTT	LLNA SI	LLNA NTT	LLNA SI	LLNA NTT					
Butyl glycidyl ether	2426-08-6	-2.51	142	130.19	AOO	10.0	25.0	50.0	—	—	1.4	2.2	5.6	—	—	31	Weak	Baskett DA et al. <sup>22</sup>	
C24 Azlactone	176664-99-6	-2.67	152	169.22	AOO	0.52	1.31	2.62	5.23	—	—	1.1	2.3	4.1	11.7	—	1.8	Moderate	Estrada E et al. <sup>22</sup>
C6 Azlactone	176665-02-4	-2.49	2.01	197.28	AOO	0.61	1.52	3.05	—	—	1.2	3.5	7.6	—	—	1.3	Moderate	Estrada E et al. <sup>22</sup>	
C9 Azlactone	176665-04-6	-2.23	2.75	239.36	AOO	1.85	3.7	7.4	—	—	1.4	4.6	10.1	—	—	2.8	Moderate	Estrada E et al. <sup>22</sup>	
Cl11 Azlactone	176665-06-8	-2.05	3.24	267.41	AOO	8.3	20.7	41.3	—	—	1.3	4	8.5	—	—	16	Weak	Estrada E et al. <sup>22</sup>	
Cl15 Azlactone	176665-09-1	-1.69	4.23	323.52	AOO	10.0	25.0	50.0	—	—	1.8	4.1	7.5	—	—	18	Weak	Estrada E et al. <sup>22</sup>	
Cl17 Azlactone	176665-11-5	-1.51	4.72	351.58	AOO	10.87	27.17	54.33	—	—	1.7	4.3	4.6	—	—	19	Weak	Estrada E et al. <sup>22</sup>	

$\Delta\text{G}^\circ = \text{acetone and olive oil (4:1)}$ ; CAS = Chemical Abstract Service number; DEP = diethyl phthalate;  $\text{dH}_2\text{O}$  = distilled water; DMF = dimethylformamide; DMSO = dimethyl sulfoxide; EtOH = ethanol; FIMA = Flavor and Extract Manufacturers' Association;  $K_{\text{ow}}$  = octanol-water partition coefficient (log scale);  $K_p$  = skin penetration coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated;

P&G = Procter & Gamble Co.; RIFM = Research Institute for Fragrance Materials.

Table I. *Continued*

Chemical Structure	CAS No.	Log K <sub>p</sub>	Log K <sub>ow</sub>	MW	Vehicule	LLNA %	LLNA EC <sub>3</sub> * (%)	Potency Category	Reference
	-1.34	5.21	379.63	AOO	11.73	29.33	58.67	—	Estrada E et al. <sup>31</sup>
	465-29-2	-2.20	2.15	166.22	AOO	5.0	10.0	25.0	—
	108-90-7	-1.85	2.19	112.56	AOO	5.0	10.0	25.0	—
	97-00-7	-4.05	-0.13	202.55	AOO	0.01	0.025	0.05	Roberts DW et al. <sup>21</sup>
	4860-03-1	-1.01	4.65	260.89	AOO	5.0	10.0	25.0	—
	26172-55-4	-2.98	0.92	149.60	DMF	0.01	0.03	0.10	Ashby J et al. <sup>21</sup>

F&G = Foger & Gamble Co.; RIFM = Research Institute for Fragrance Materials.

\*Mathematically estimated.  
†Value estimated.

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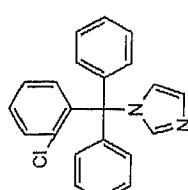
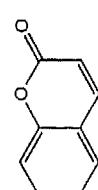
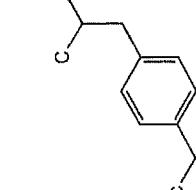
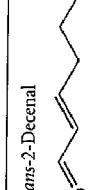
Table 1. *Continued*

acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMSO = dimethyl sulfoxide; EtOH = ethanol; FEMA = Flavor and Extract Manufacturers' Association; K<sub>ow</sub> = octanol-water partition coefficient (log scale); K<sub>p</sub> = skin penetration coefficient (log scale); LINA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated;

&G = Procter & Gamble Co.; RIFM = Research Institute for Fragrance Materials.

Mathematically est

Table I. Continued

Chemical Structure	CAS No.	Log K <sub>ow</sub>	MW	Vehicle	LLNA EC <sub>3*</sub> (%)	Category	Potency	Reference											
	23593-75-1	-1.02	5.35	344.85	AOO	2.5	5.0	10.0	—	—	1.6	3.1	3.0	—	—	4.8	Moderate	Estrada E et al. <sup>22</sup>	
	91-64-5	-2.26	1.91	146.15	AOO	5.0	10.0	25.0	—	—	2.7	2.9	2.3	—	—	NC	Nonsensitizer	RIFM/FEMA database	
	103-95-7	-1.55	3.28	190.29	AOO	1.0	2.5	10.0	25.0	50.0	—	1.4	1.3	1.8	3.3	5.2	—	Weak	Baskett DA et al. <sup>16</sup>
	3913-71-1	-1.86	2.54	154.25	AOO	0.5	1.0	2.5	5.0	10.0	—	1.3	1.1	3	6	9.5	—	Moderate	Patlewitz G et al. <sup>34</sup>
	35691-65-7	-2.99	1.91	265.94	AOO	0.5	1.0	2.5	5.0	—	1.4	3.4	3.5	5.4	—	0.9	Strong	Unilever (unpublished)	

AOO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMF = dimethylformamide; DMSO = dimethyl sulfoxide; EtOH = ethanol; FEMA = Flavor and Extract Manufacturers' Association; K<sub>ow</sub> = octanol-water partition coefficient (log scale); K<sub>p</sub> = skin penetration coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated; P&G = Procter & Gamble Co.; RIFM = Research Institute for Fragrance Materials.

\*Mathematically estimated concentration of test chemical necessary to induce a threefold stimulation index (SI).

(continued on next page)

Table I. Continued

Chemical Structure	CAS No.	$\log K_{ow}$	MW	Vehicle	LLNA %	LLNA %	LLNA %	LLNA %	LLNA %	LLNA %	LLNA %	LLNA %	LLNA EC <sub>50</sub> (%)	Category	Potency	Reference			
Diethyl acetaldehyde	97-96-1	-2.23	1.56	100.16	AOO	25.0	50.0	75.0	100.0	—	—	1.2	0.8	2.4	16.3	—	76	Weak	Paluszewicz G et al. <sup>13</sup>
Dithylenetriamine	111-40-0	-3.14	0.29	103.17	AOO	10.0	25.0	—	—	—	—	6.4	12.1	—	—	—	5.8	Moderate	Baskettier DA et al. <sup>12</sup>
Dithiyl maleate	141-05-9	-3.14	0.89	172.18	AOO	25.0	50.0	100.0	—	—	—	16.3	22.6	13.1	—	—	5.8 <sup>†</sup>	Moderate	Ryan CA et al. <sup>18</sup>
1-(25-Diethylphenyl)butane-1,3-dione	167998-76-7	-1.84	3.14	221.32	Acetone	10.0	20.0	40.0	—	—	—	3.9	19.2	18.7	—	—	9.6 <sup>†</sup>	Moderate	P&G (unpublished)
Diethylphthalate	84-66-2	-2.75	1.87	222.24	AOO	25.0	50.0	100.0	—	—	—	1.0	1.3	1.5	—	—	NC	Nonsensitizer	Ryan CA et al. <sup>12</sup>

AOO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMF = dimethylformamide; DMSO = dimethyl sulfoxide; EtOH = ethanol; FEMAA = Flavor and Extract Manufacturers' Association;  $K_{ow}$  = octanol-water partition coefficient (log scale); Kp = skin penetration coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated;

P&G = Procter & Gamble Co.; RIFM = Research Institute for Fragrance Materials.

\*Mathematically estimated concentration of test chemical necessary to induce a threefold stimulation index (SI).

<sup>†</sup>Value estimated.

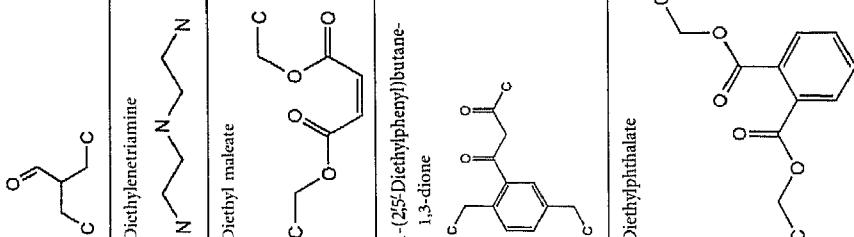
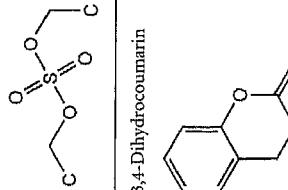
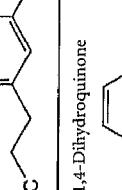
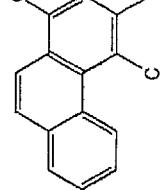


Table 1. Continued

Chemical Structure	CAS No.	Log K <sub>ow</sub>	Log K <sub>p</sub>	MW	Vehicle	LLNA %	LLNA %	LLNA %	LLNA %	LLNA EC <sub>3*</sub> (%)	Potency Category	Reference
Dietyl sulfate	64-67-5	-3.73	-0.09	154.18	AOO	1.0	2.5	10.0	—	—	Moderate	Ashby J et al. <sup>21</sup>
	119-84-6	-2.27	1.91	148.16	AOO	2.5	5.0	10.0	—	—	Moderate	Ashby J et al. <sup>21</sup>
Dihydrogenol	2785-87-7	-2.20	2.15	166.22	AOO	5.1	10.1	25.3	—	—	Moderate	Smith CK, Hotchkiss SA. <sup>27</sup>
	123-31-9	-2.56	1.17	110.11	AOO	0.1	0.25	0.5	1.0	2.5	—	Kimber I et al. <sup>35</sup>
3-Dimethylaminopropylamine	109-55-7	-2.69	0.92	102.18	AOO	0.5	1.0	2.5	5.0	10.0	—	Wright ZM et al. <sup>36</sup>
	57-97-6	-0.46	5.39	256.35	DMF	0.025	0.5	1.0	—	—	Extreme	Ashby J et al. <sup>21</sup>

AOO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMF = dimethylformamide; DMSO = dimethyl sulfoxide; EtOH = ethanol; K<sub>ow</sub> = octanol-water partition coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated; Manufacturers' Association; P&G = Procter & Gamble Co.; RIFM = Research Institute for Fragrance Materials.

\*Mathematically estimated concentration of test chemical necessary to induce a threefold stimulation index (SI).

<sup>t</sup>Value estimated.

(continued on next page)

Table 1. Continued

Chemical Structure	CAS No.	$\log K_{ow}$	$\log K_p$	MW	Vehicle	LLNA %	LLNA EC <sub>3*</sub> (%)	Potency Category	Reference										
5,5-Dimethyl-3-methylene-dihydro-2(3H)-furanone	29043-97-8	-2.48	1.42	126.16	AOO	2.0	4.0	8.0	—	—	3	7.4	9.2	—	—	1.8 <sup>†</sup>	Moderate	Ashby J et al. <sup>21</sup>	
1-(2 <sup>5</sup> S-Dimethylphenyl)butane-1,3-dione	56290-55-2	-2.02	2.65	193.27	Acetone	10.0	20.0	40.0	—	—	—	2.3	5.1	9.5	—	—	13	Weak	P&G (unpublished)
Dimethyl sulfate	77-78-1	-3.91	-0.59	126.13	AOO	0.25	0.5	1.0	—	—	—	3.8	6.0	5.7	—	—	0.19 <sup>†</sup>	Strong	Ashby J et al. <sup>21</sup>
Dimethylsulfoxide	67-68-5	-2.79	0.57	78.13	AOO	25.0	50.0	100.0	—	—	—	2.7	2.3	3.9	—	—	72	Weak	Estrada E et al. <sup>22</sup>
Dodecyl methane sulfonate	51323-71-8	-2.55	2.51	264.42	AOO	5.0	10.0	25.0	—	—	—	2.1	3.3	9	—	—	8.8	Moderate	Ashby J et al. <sup>21</sup>

AOO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMF = dimethylformamide; DMSO = dimethyl sulfoxide; EtOH = ethanol; FEMA = Flavor and Extract Manufacturers' Association;  $K_{ow}$  = octanol-water partition coefficient (log scale);  $K_p$  = skin penetration coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated; P&G = Procter & Gamble Co.; RIEM = Research Institute for Fragrance Materials.

\*Mathematically estimated concentration of test chemical necessary to induce a threefold stimulation index (SI).

<sup>†</sup>Value estimated.

Table 1. Continued

Chemical Structure	CAS No.	Log K <sub>p</sub>	Log K <sub>OW</sub>	MW	Vehicle	LLNA %	Patency Category	Reference												
	15646-46-5	-2.71	1.87	217.22	AOO	0.0025	0.005	0.01	0.025	0.05	—	2.9	4.9	12.0	22.0	33.0	—	0.003	Extreme	Loveless SE et al. <sup>33</sup>
	170928-69-5	-2.11	3.00	248.32	Acetone	10.0	20.0	40.0	—	—	—	1.1	1.7	3.7	—	—	33	Weak	P&G (unpublished)	
	140-88-5	-2.67	0.92	100.12	AOO	10.0	25.0	50.0	—	—	—	1.2	2.7	5	—	—	28	Weak	Warbrick EV et al. <sup>29</sup>	
	94-02-0	-2.46	2.01	192.21	Acetone	10.0	20.0	40.0	—	—	—	0.9	0.9	1.2	—	—	NC	Nonsensitizer	P&G (unpublished)	
	107-15-3	-2.95	0.19	60.10	AOO	0.1	0.25	0.5	1.0	2.5	5.0	1.1	1.2	1.6	1.9	3.3	6.1	2.2	Moderate	Kimber I et al. <sup>35</sup>

AOO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMF = dimethylformamide; DMSO = dimethyl sulfoxide; EtOH = ethanol; FFM = Flavor and Extract Manufacturers' Association; K<sub>p</sub>/K<sub>OW</sub> = octanol-water partition coefficient (log scale); K<sub>p</sub> = skin penetration coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated; P&G = Procter & Gamble Co.; RIFM = Research Institute for Fragrance Materials.

\* Mathematically estimated concentration of test chemical necessary to induce a threefold stimulation index (SI).

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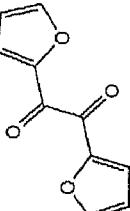
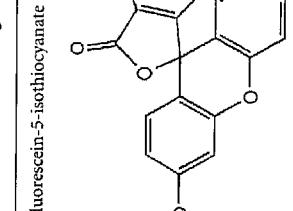
Table 1. Continued

Chemical Structure	CAS No.	Log K <sub>ow</sub>	Log K <sub>p</sub>	MW	Vehicle	% LLNA	% LLNA %	% LLNA %	% LLNA %	% LLNA	% LLNA	Potency Category	LLNA EC <sub>50</sub> * (%)	Reference						
Ethylene glycol dimethacrylate	97-90-5	-2.95	1.38	198.22	MEK	10.0	25.0	50.0	—	—	1.2	2.4	7	—	—	—	28	Weak	Unilever (unpublished)	
4-(N-Ethyl-N-(2-methanesulphonamidoethyl)-2-methyl-1,4-phenylenediamine (CD3))	25646-71-3	-9.33	-2.12	836.97	DMSO	0.1	1.0	5.0	10.0	—	—	1.2	4.5	5.9	6.3	—	—	0.6	Strong	Ryan CA et al. <sup>28</sup>
N-Ethyl-N-nitrosourea	759-73-9	-3.95	-0.73	117.11	AOO	0.25	1.0	10.0	—	—	—	—	—	—	—	—	—	—	Ashtby J et al. <sup>21</sup>	
Ethyl vanillin	121-32-4	-2.65	1.52	166.18	AOO	2.5	5.0	10.0	25.0	50.0	—	0.65	1.05	0.74	0.36	0.29	—	NC	Nonsensitizer	Baskett DA et al. <sup>16</sup>

AOO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMF = dimethylformamide; DMSO = dimethyl sulfoxide; EtOH = ethanol; FEMA = Flavor and Extract Manufacturers' Association; K<sub>ow</sub> = octanol-water partition coefficient (log scale); K<sub>p</sub> = skin penetration coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated; P&G = Procter & Gamble Co.; RIFM = Research Institute for Fragrance Materials.

\*Mathematically estimated concentration of test chemical necessary to induce a threefold stimulation index (SI).

Table 1. Continued

Chemical Structure	CAS No.	Log K <sub>ow</sub>	MW	Vehicle	% VNTT	LLNA %	LLNA EC <sub>3*</sub> (%)	Potency Category	Reference											
Eugenol	97-53-0	-2.19	215	164.20	AOO	2.5	5.0	10.0	25.0	—	1.6	1.5	2.4	5.5	—	13	Weak	Lovelace SE et al. <sup>33</sup>		
Farnesol	502-67-0	-1.39	3.77	220.36	AOO	1.0	2.5	5.0	10.0	25.0	—	0.6	1.1	1.7	2.5	7	—	12	Weak	Patlewicz G et al. <sup>34</sup>
	3326-32-7	-2.74	3.32	389.38	Acetone/dibutyl phthalate, 50/50	0.5	1.0	2.5	—	—	—	8.6	11.7	16.6	—	—	0.14 <sup>i</sup>	Strong	Ashby J et al. <sup>21</sup>	
	50-00-0	-2.67	0.33	30.03	Acetone	0.093	0.185	0.37	0.925	1.85	—	1.1	2.3	2.3	3.9	4.0	—	0.61	Strong	Hilton J et al. <sup>35</sup>
	492-94-4	-2.90	1.38	190.15	AOO	5.0	10.0	25.0	—	—	—	1.2	1.7	2.2	—	—	NC	Nonsensitizer	Roberts DW et al. <sup>31</sup>	

AOO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMF = dimethylformamide; DMSO = dimethyl sulfoxide; EtOH = ethanol; FEMA = Flavor and Extract Manufacturers' Association; K<sub>ow</sub> = octanol-water partition coefficient (log scale); K<sub>p</sub> = skin penetration coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated; P&G = Procter & Gamble Co.; RIEM = Research Institute for Fragrance Materials.

\*Mathematically estimated concentration of test chemical necessary to induce a threefold stimulation index (SI).

<sup>i</sup>Value estimated.

<sup>t</sup>Concentration shown as actual concentration of formaldehyde. Material tested was formalin (37% formaldehyde).

(continued on next page)

Table 1. *Continued*

AAOO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMSO = dimethylformamide; DMF = dimethyl sulfoxide; EtOH = ethanol; FEMA = Flavor and Extract Manufacturers' Association;  $K_{ow}$  = octanol-water partition coefficient (log scale);  $K_p$  = skin penetration coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated;

Manufacturers Association, KOW = Octanoic acid, parton compound,  $\omega_{\text{parton}} = \frac{\text{parton}}{\text{parton} + \text{KOW}}$ , RIFM = Research Institute for Fragrance Materials.

Table 1. Continued

Chemical Structure	CAS No.	Log K <sub>p</sub>	MW	Vehicle	LLNA %	LLNA EC <sub>3</sub> * (%)	Category	Potency	Reference	
4-Hydrobenzoic acid	99-96-7	-2.83	1.03	138.12	DMSO	5.0	10.0	25.0	—	—
Hydroxycitronellal	107-75-5	-2.24	2.15	172.27	AOO	2.5	5.0	10.0	25.0	50.0
2-Hydroxyethyl acrylate	818-61-1	-3.05	0.54	116.12	AOO	5.0	10.0	25.0	—	—
3 and 4-(4-Hydroxy-4-methylphenyl)-3-cyclohexene-1-carboxaldehyde (Lyral)	31906-04-4	-1.95	2.89	210.32	AOO	1.0	2.5	5.0	10.0	25.0
2-Hydroxypropyl methacrylate	923-26-2	-2.87	1.03	144.17	AOO	10.0	25.0	50.0	—	—
Imidazolidinylurea	39236-46-9	-7.22	-3.00	388.30	DMF	10.0	25.0	50.0	—	—

AOO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMF = dimethylformamide; DMSO = dimethyl sulfoxide; EtOH = ethanol; FEMA = Flavor and Extract Manufacturers' Association; Kow = octanol-water partition coefficient (log scale); K<sub>p</sub> = skin penetration coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated; P&G = Procter & Gamble Co.; RIFM = Research Institute for Fragrance Materials.

\*Mathematically estimated concentration of test chemical necessary to induce a threefold stimulation index (SI).

<sup>a</sup>Value estimated.

(continued on next page)

Table 1 (Continued)

Chemical Structure	UV-vis		Vis-NIR		LUMO		HOMO		PCP		PCP*		Reference					
	λ <sub>max</sub> (nm)	ε <sub>max</sub>																
1-Iodoheptane	344.77-41.39	4.89	352.35	AOO	10.0	25.0	50.0	—	—	1.6	3.9	6.4	—	19	Weak	Estrada E et al. <sup>22</sup>		
1-Iodohexadecane	638.45-5.9	2.29	2.43	212.08	AOO	10.0	25.0	50.0	—	—	0.9	1.2	2.5	—	—	NC	Nonsensitizer Unilever (unpublished)	
1-Iodohexane	4282.42-2	-2.02	3.17	254.16	AOO	10.0	25.0	50.0	—	—	1.3	3.1	4.6	—	—	24	Weak	Estrada E et al. <sup>22</sup>
1-Iodooctadecane	629.93-6	-1.22	5.39	380.40	AOO	5.0	10.0	25.0	—	—	1	1.4	1.9	—	—	NC	Nonsensitizer Unilever (unpublished)	
1-Iodotetradecane	19218.94-1	-1.57	4.40	324.29	AOO	10.0	25.0	50.0	—	—	1.7	6.9	9.7	—	—	14	Weak	Ashby J et al. <sup>21</sup>
Isoeugenol	97.54-1	-2.19	2.15	164.20	AOO	0.50	1.0	5.0	—	—	1	1.1	12.4	—	—	1.2	Moderate	Baskett DA, Caddy P. <sup>41</sup>
[Isobornanoyl] chloride	57077-36-8	-1.99	2.54	176.69	AOO	5.0	10.0	25.0	—	—	6.6	10.6	12.6	—	—	2.7 <sup>f</sup>	Moderate	Unilever (unpublished)
Isopropanol	67-63-0	-2.51	0.82	60.10	AOO	10.0	25.0	50.0	—	—	1.7	1.1	1.0	—	—	NC	Nonsensitizer Baskett DA et al. <sup>39</sup>	

acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMSO = dimethyl sulfoxide; EtOH = ethanol; FEMA = Flavor and Extract Manufacturers' Association; Kow = octanol-water partition coefficient (log scale); K<sub>p</sub> = skin penetration coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated;

the same substances can act as both inducers and inhibitors of angiogenesis.

Table 1. Continued

Chemical Structure	CAS No.	$\log K_{ow}$	Vehicle	MW	LLNA %	LLNA EC <sub>3*</sub> (%)	LLNA Category	Potency	Reference
	51474-90-9	-1.93	2.89	206.29	AOO	12.0	29.0	59.0	—
Isopropyl eugenol								1.8	2.2
								—	—
								NC	Nonsensitizer
									Bertrand F et al. <sup>42</sup>
	2953-00-7	-1.93	2.89	206.29	AOO	0.6	1.2	3.0	—
Isopropyl isobutugenol								—	—
								3	5.7
								10 <sup>7</sup>	—
								—	0.6 <sup>†</sup>
									Strong
									Bertrand F et al. <sup>42</sup>
	110-27-0	-1.62	3.88	270.46	AOO	25.0	50.0	100.0	—
Isopropyl myristate								—	—
								2.1	3.3
								3.4	—
								—	44
									Ryan CA et al. <sup>38</sup>
	59-01-8; 8063-07-8	-6.31	-0.90	484.50	AOO	5.0	10.0	25.0	—
Kanamycin								—	—
								2.2	0.8
								1	—
								—	—
								NC	Nonsensitizer
									Baskett DA et al. <sup>39</sup>
	50-21-5	-3.24	0.05	90.08	DMSO	5.0	10.0	25.0	—
Lactic acid								—	—
								1.0	1.4
								2.2	—
								—	NC
									Nonsensitizer
									Baskett DA et al. <sup>39</sup>

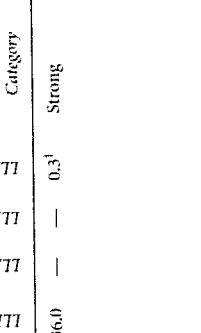
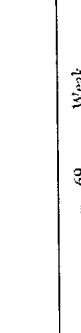
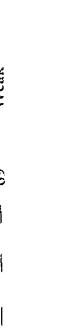
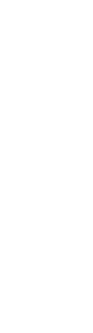
\*Mathematically estimated concentration of test chemical necessary to induce a threefold stimulation index (SI).

<sup>†</sup>Value estimated.

(continued on next page)

AOO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMF = dimethylformamide; DMSO = dimethyl sulfoxide; EOIH = ethanol; FEMA = Flavor and Extract Manufacturers' Association;  $K_{ow}$  = octanol-water partition coefficient (log scale);  $K_p$  = skin penetration coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated; P&G = Procter & Gamble Co.; RIEM = Research Institute for Fragrance Materials.

Table 1. Continued

Chemical Structure	CAS No.	Log K <sub>ow</sub>	MW	Vehicle	LLNA %	LLNA SI	LLNA %	LLNA SI	LLNA %	LLNA SI	LLNA %	LLNA %	LLNA SI	LLNA %	LLNA %	LLNA EC <sub>50</sub> (μg/ml)	Potency Category	Reference	
	1166-52-5	-2.51	3.21	338.44	DMSO	1.0	10.0	25.0	50.0	—	—	12.1	29.7	29.3	36.0	—	0.3 <sup>1</sup>	Strong	P&G (unpublished)
	5989-27-5	-1.47	2.93	136.24	AOO	25.0	50.0	100.0	—	—	—	1.8	2.4	4.0	—	—	69	Weak	Warbrick EV et al. <sup>29</sup>
	78-70-6	-1.86	2.54	154.25	AOO	25.0	50.0	100.0	—	—	—	2.5	4.8	8.3	—	—	30	Weak	Ryan CA et al. <sup>28</sup>
	149-30-4	-2.46	1.80	167.24	DMF	1.0	3.0	10.0	—	—	—	2.3	4.4	8.6	—	—	1.7	Moderate	Baskettet DA et al. <sup>44</sup>
	100-06-1	-2.28	1.91	150.18	AOO	10.0	25.0	50.0	—	—	—	1.3	1.0	1.0	—	—	NC	Nonsensitizer	Ryan CA et al. <sup>28</sup>

AOO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMSO = dimethyl sulfoxide; EtOH = ethanol; FMEA = Flavor and Extract Manufacturers' Association; K<sub>ow</sub> = octanol-water partition coefficient (log scale); K<sub>p</sub> = skin penetration coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated;

PP&G = Procter & Gamble Co.; RIFM = Research Institute for Fragrance Materials.

Mathematically educated

Table 1. *Continued*

<sup>a</sup>Q = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMF = dimethylformamide; DMSO = dimethyl sulfoxide; EtOH = ethanol; FEMa = Flavor and Extract Manufacturers' Association;  $K_{ow}$  = octanol-water partition coefficient (log scale); K<sub>p</sub> = skin penetration coefficient (log scale); LINa = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated; G = Procter & Gamble Co.; RIFM = Research Institute for Fragrance Materials.

mathematically estimated.

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Table 1. Continued

Chemical Structure	CAS No.	$\log K_{ow}$	MW	Vehicle	LLNA %	LLNA EC3* (%)	Potency Category	Reference											
	2374-65-4	-2.55	251	264.42	AOO	1.0	2.5	5.0	—	—	21.6	39.9	48.6	—	—	0.39 <sup>†</sup>	Strong	Bakkeret DA, Scholles EW. <sup>26</sup>	
	95-70-5	-2.46	1,416	122.08	DMSO	0.125	0.25	0.5	1.0	—	—	2.6	3.5	4.1	5.5	—	0.2	Strong	P&G (internal)
	186743-26-0	-2.10	2.40	178.23	AOO	11.0	27.0	54.0	—	—	—	1.5	2.3	6.4	—	—	3.2	Weak	Bertrand F et al. <sup>42</sup>
	186743-25-9	-2.10	2.40	178.23	AOO	11.0	27.0	54.0	—	—	—	2.7	4.9	4.3	—	—	13	Weak	Bertrand F et al. <sup>42</sup>
	186743-24-8	-2.10	2.40	178.23	AOO	11.0	27.0	54.0	—	—	—	1.9	4.9	8.3	—	—	17	Weak	Bertrand F et al. <sup>42</sup>

AOO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMF = dimethylformamide; DMSO = dimethyl sulfoxide; EOIH = ethano; FEMA = Flavor and Extract Manufacturers' Association; K<sub>ow</sub> = octanol-water partition coefficient (log scale); K<sub>p</sub> = skin penetration coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated; P&G = Procter & Gamble Co.; RIFM = Research Institute for Fragrance Materials.

\*Mathematically estimated concentration of test chemical necessary to induce a threefold stimulation index (S1).

<sup>†</sup>Value estimated.

Table I. Continued

Chemical Structure	CAS No.	$\log K_{ow}$	$\log K_p$	MW	Vehide	LLNA %	LLNA %	LLNA %	LLNA %	LLNA EC <sub>50</sub> (%)	Potency Category	Reference
Methyl hexadecene sulfonate	26452-48-2	-2.19	3.49	318.52	AOO	5.0	10.0	25.0	—	—	IS VNTI	Ashby J et al. <sup>21</sup>
Methyl hexadecyl sulfonate	4230-15-3	-2.20	3.49	320.53	AOO	5.0	10.0	25.0	—	—	IS VNTI	(unpublished)
5-Methyl-2,3-hexanedione	13706-86-0	-2.50	1.42	128.17	AOO	25.0	50.0	100.0	—	—	IS VNTI	Ryan CA et al. <sup>28</sup>
p-Methylhydrocinnamic aldehyde	5406-12-2	-1.82	2.54	148.21	AOO	2.5	5.0	10.0	25.0	50.0	IS VNTI	Baskett DA et al. <sup>16</sup>
Methyl 4-hydroxybenzoate (methylparaben)	99-76-3	-2.74	1.28	152.15	DMF	10.0	25.0	50.0	—	—	IS VNTI	Ryan CA et al. <sup>28</sup>
2-Methyl-5-hydroxyethylaminophenol	55302-96-0	-2.66	1.52	167.21	AOO	0.1	0.25	0.5	1.0	2.5	IS VNTI	Estrada E et al. <sup>22</sup>

AOO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMF = dimethylformamide; DMSO = dimethyl sulfoxide; EtOH = ethanol; FEMA = Flavor and Extract Manufacturers' Association; K<sub>ow</sub> = octanol-water partition coefficient (log scale); K<sub>p</sub> = skin penetration coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated; \*Mathematically estimated concentration of test chemical necessary to induce a threefold stimulation index (SI).

<sup>t</sup>Value estimated.

(continued on next page)

Table 1. *Continued*

AOO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMSO = dimethylformamide; DMF = dimethyl sulfoxide; EtOH = ethanol; FEMA = Flavor and Extract Manufacturers' Association; K<sub>ow</sub> = octanol-water partition coefficient (log scale); K<sub>p</sub> = skin penetration coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated;

International Fragrances Association, RIFM = Research Institute for Fragrance Materials.  
P&G = Procter & Gamble Co.; RIWF = Research Institute for Fragrance Materials.

Table 1. Continued

Chemical Structure	CAS No.	Log K <sub>p</sub>	Log K <sub>ow</sub>	Vehicle	% VNA	% LLNA	% VNTT	% LLNA EC3*	Potency Category	Reference
N-Methyl-N-nitrosourea	684-93-5	-4.04	-0.97	103.08	AOO	0.05	0.1	0.25	—	—
										Ashby J et al. <sup>21</sup>
Methyl 2-nonyoate	111-80-8	-2.22	2.15	168.24	80%	5.0	10.0	20.0	—	—
				EtOH						Ryan CA et al. <sup>28</sup>
$\alpha$ -Methylphenylacetaldehyde	93-53-8	-1.91	2.29	134.18	AOO	0.5	1.0	2.5	5.0	10.0
								—	2	Patlewitz G et al. <sup>24</sup>
3-Methyl-4-phenyl-1,2,5-thiadiazole-1,1-dioxide (MPT)	3775-21-1	-3.18	1.14	208.24	AOO	0.1	0.25	0.5	1.0	2.5
								—	1.3	Estrada E et al. <sup>22</sup>
Methyl salicylate	119-36-8	-2.74	1.28	152.15	AOO	1.0	2.5	5.0	10.0	20.0
								—	1.0	Kimber I et al. <sup>15</sup>

AOO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMF = dimethylformamide; DMSO = dimethyl sulfoxide; EtOH = ethanol; FEMa = Flavor and Extract Manufacturers' Association; K<sub>ow</sub> = octanol-water partition coefficient (log scale); K<sub>p</sub> = skin penetration coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated; P&G = Procter & Gamble Co.; RIFM = Research Institute for Fragrance Materials.

\*Mathematically estimated concentration of test chemical necessary to induce a threefold stimulation index (SI).

<sup>†</sup>Value estimated.

(continued on next page)

Table 1. *Continued*

Manufacturers Association, NOW - ocean-air products, now - g. P&G = Procter & Gamble Co.; RIEM = Research Institute for Fragrance Materials.

\*Mathematically estimated concentration of test chemical necessary to induce a targeted stimulus.

Table 1. Continued

Chemical Structure	CAS No.	$\log K_{ow}$	$\log K_p$	MW	Vehicile	LLNA	LLNA EC3*	Potency Category	Reference
Nonanoyl chloride	764-85-2	-1.99	2.54	176.69	AOO	5.0	10.0	25.0	IS VNTT
cis-6-Nonenal	2277-19-2	-1.95	2.29	140.23	AOO	10.0	25.0	50.0	IS VNTT
Octanoic acid	124-07-2	-2.42	1.66	144.21	AOO	10.0	25.0	50.0	IS VNTT
Oleyl methane sulfonate	35709-09-2	-2.01	3.98	346.57	AOO	5.0	10.0	25.0	IS VNTT
Oxalic acid	144-62-7	-3.69	-0.59	90.03	DMF	5.0	10.0	25.0	IS VNTT
Palmitoyl chloride	112-67-4	-1.37	4.26	274.88	AOO	5.0	10.0	25.0	IS VNTT
LLNA % NNTT									
cis-6-Nonenal									
Octanoic acid									
Oleyl methane sulfonate									
Oxalic acid									
Palmitoyl chloride									
LLNA % VNTT									
Octanoic acid									
Oleyl methane sulfonate									
Oxalic acid									
Palmitoyl chloride									
LLNA % NNTT									
cis-6-Nonenal									
Octanoic acid									
Oleyl methane sulfonate									
Oxalic acid									
Palmitoyl chloride									
LLNA % VNTT									
Octanoic acid									
Oleyl methane sulfonate									
Oxalic acid									
Palmitoyl chloride									
LLNA % NNTT									
cis-6-Nonenal									
Octanoic acid									
Oleyl methane sulfonate									
Oxalic acid									
Palmitoyl chloride									
LLNA % VNTT									
Octanoic acid									
Oleyl methane sulfonate									
Oxalic acid									
Palmitoyl chloride									
LLNA % NNTT									
cis-6-Nonenal									
Octanoic acid									
Oleyl methane sulfonate									
Oxalic acid									
Palmitoyl chloride									
LLNA % VNTT									
Octanoic acid									
Oleyl methane sulfonate									
Oxalic acid									
Palmitoyl chloride									
LLNA % NNTT									
cis-6-Nonenal									
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Oleyl methane sulfonate									
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Oxalic acid									
Palmitoyl chloride									
LLNA % VNTT									
Octanoic acid									
Oleyl methane sulfonate									
Oxalic acid									
Palmitoyl chloride									
LLNA % NNTT									
cis-6-Nonenal									
Octanoic acid									
Oleyl methane sulfonate									
Oxalic acid									

Table 1. *Continued*

AOO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DME = dimethylformamide; DMSO = dimethyl sulfoxide; EtOH = ethanol; FEMA = Flavor and Extract Manufacturers' Association; *K*<sub>ow</sub> = octanol-water partition coefficient (log scale); *K*<sub>p</sub> = skin penetration coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated;

P&G = Procter & Gamble Co.; RIFM = Research Institute for Fragrance Materials.  
\*Mathematically estimated concentration of test chemical necessary to induce a threefold stimulation index (SI).

Table 1. *Continued*

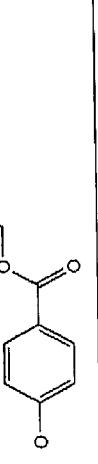
OO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMF = dimethylformamide; DMSO = dimethyl sulfoxide; EtOH = ethanol; FEMA = Flavor and Extract Manufacturers' Association; K<sub>ow</sub> = octanol/water partition coefficient (log scale); K<sub>p</sub> = skin penetration coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated;

P&G = Procter & Gamble Co.; RIFM = Research Institute for Fragrance Materials.

Mathematically estimated  
Value estimated

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Table 1. Continued

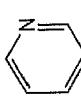
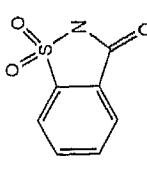
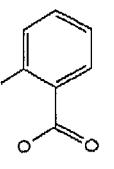
Chemical Structure	CAS #	Log K <sub>ow</sub>	Log K <sub>p</sub>	MW	Vehicle	LLNA %	LLNA SI	LLNA SI	LLNA SI	LLNA SI	LLNA EC <sub>3*</sub> (%)	Potency Category	Reference	
3-Phenyl propenal	14371-10-9	-1.90	2.29	132.16	AOO	1.0	2.5	5.0	10.0	25.0	—	2.4	Moderate	Patlewitz G et al. <sup>14</sup>
Potassium dichromate	7778-50-9	-6.10	-2.24	294.18	DMSO	0.025	0.05	0.1	0.25	0.5	—	1.6	1.4	Kimber I et al. <sup>17</sup>
														
$\beta$ -Propiolactone	57-57-8	-2.85	0.43	72.06	AOO	0.025	1.0	2.5	—	—	1.5	13.0	Strong	Ashby J et al. <sup>21</sup>
														
Propylene glycol	57-55-6	-2.88	0.43	76.10	dH <sub>2</sub> O	50.0	100.0	—	—	—	1.2	1.6	—	NC
													Baskett DA et al. <sup>19</sup>	
3-Propyldenephthalide	17369-59-4	-2.08	2.40	174.20	AOO	5.0	10.0	20.0	—	—	4.9	9.1	3.7 <sup>i</sup>	Ryan CA et al. <sup>28</sup>
														
Propylparaben	94-13-3	-2.56	1.77	180.20	AOO	5.0	10.0	25.0	—	—	1.4	1.0	—	NC
													Ashby J et al. <sup>21</sup>	

AOO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMSO = dimethylformamide; EtOH = ethanol; FEMA = Flavor and Extract Manufacturers' Association; K<sub>ow</sub> = octanol-water partition coefficient (log scale); K<sub>p</sub> = skin penetration coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated; P&G = Procter & Gamble Co.; RJFM = Research Institute for Fragrance Materials.

\*Mathematically estimated concentration of test chemical necessary to induce a threefold stimulation index (SI).

<sup>i</sup>Value estimated.

Table 1. Continued

Chemical Structure	CAS No.	Log K <sub>ow</sub>	Vehicle	LLNA %	LLNA %	LLNA EC <sub>3</sub> * (%)	Potency Category	Reference											
Pyridine	110-86-1	-2.27	1.31	79.10	AOO	25.0	50.0	100.0	—	1.1	2.3	3.9	—	—	72	Weak	Baskett DA et al. <sup>32</sup>		
																			
Resorcinol	108-46-3	-2.56	1.17	110.11	DMF	5.0	10.0	25.0	—	—	—	2.2	2.7	—	—	NC	Nonsensitizer	Warbrick EV et al. <sup>33</sup>	
																			
Saccharin	81-07-2	-3.38	0.64	183.18	DMSO	25.0	50.0	75.0	—	—	—	1.3	1.5	1.5	—	—	Nonsensitizer	Warbrick EV et al. <sup>33</sup>	
																			
Salicylic acid	69-72-7	-2.83	1.03	138.12	AOO	5.0	10.0	25.0	—	—	—	0.8	1.5	2.5	—	—	NC	Nonsensitizer	Baskett DA et al. <sup>33</sup>
																			
Sodium lauryl sulfate	151-21-3	-3.15	1.87	288.38	DMF	1.0	2.5	5.0	10.0	20.0	—	0.9	1.1	1.7	2.6	3.5	—	14.0	Weak (false positive)
																			

AOO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMF = dimethyl formamide; DMSO = dimethyl sulfoxide; EtOH = ethanol; FEMIA = Flavor and Extract Manufacturers' Association; K<sub>ow</sub> = octanol-water partition coefficient (log scale); K<sub>p</sub> = skin penetration coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated; P&G = Procter & Gamble Co.; RIFM = Research Institute for Fragrance Materials.

\*Mathematically estimated concentration of test chemical necessary to induce a threefold stimulation index (SI).

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Table 1. Continued

Chemical Structure	CAS No. <sup>a</sup>	Log K <sub>p</sub>	Log K <sub>ow</sub>	MW	Vehicle	LLNA %	LLNA EC3* (%)	Potency Category	Reference											
Sodium 3,5,5'-trimethylhexanoyloxybenzenesulfonate	94612-91-6	-3.19	2.23	336.38	DMSO	5.0	10.0	25.0	—	—	2.3	4.8	7.8	—	—	6.4	Moderate	Ashby J et al. <sup>21</sup>		
Na <sup>+</sup>																				
Streptomycin sulfate	3810-74-0	-17.65	-8.50	1457.38	DMF	2.5	5.0	10.0	25.0	50.0	—	1.2	1.4	1.3	2.0	1.9	—	NC	Nonsensitizer	Kimber L et al. <sup>35</sup>
Sulfanilamide	63-74-1	-3.49	0.40	172.20	DMF	10.0	25.0	50.0	—	—	—	1	1	0.9	—	—	—	NC	Nonsensitizer	Baskett DA et al. <sup>32</sup>

AOO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMF = dimethylformamide; DMSO = dimethyl sulfoxide; EtOH = ethanol; FEMA = Flavor and Extract Manufacturers' Association; K<sub>ow</sub> = octanol-water partition coefficient (log scale); K<sub>p</sub> = skin penetration coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated; P&G = Procter & Gamble Co.; RIFM = Research Institute for Fragrance Materials.

\*Mathematically estimated concentration of test chemical necessary to induce a threefold stimulation index (SI).

Table 1. Continued

Chemical Structure	CAS No.	Log K <sub>p</sub>	Log K <sub>ow</sub>	MW	Vehicle	LLNA %	LLNA EC <sub>3</sub> * (%)	Potency Category	Reference									
Sulfamilic acid	121-57-3	-3.49	0.40	173.19	DMF	5.0	10.0	25.0	—	—	1.5	1.9	2.2	—	—	NC	Nonsensitizer	Baskett DA et al. <sup>48</sup>
Tartaric acid	87-69-4	-4.25	-0.87	150.09	DMF	5.0	10.0	25.0	—	—	1.6	3.5	6.7	—	—	8.7	Moderate	Unilever (unpublished)
Tetrachlorosalicylanilide	1154-59-2	-2.38	3.49	351.02	Acetone	0.25	0.5	1.0	—	—	11.2	14.4	18.0	—	—	0.04 <sup>†</sup>	Extreme	Baskett DA et al. <sup>32</sup>
2,2,6,6-Tetramethyl-heptane-3,5-dione	1118-71-4	-2.15	2.40	186.30	Acetone	10.0	20.0	40.0	—	—	2.1	2.8	3.4	—	—	27	Weak	P&G (unpublished)

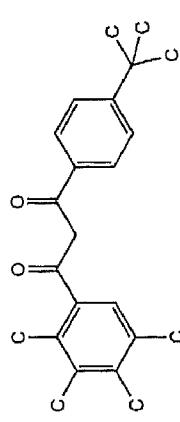
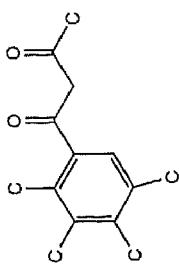
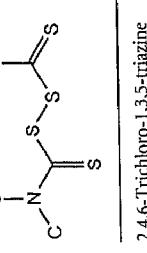
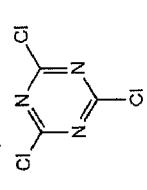
AOO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMF = dimethylformamide; DMSO = dimethyl sulfoxide; EtOH = ethanol; FEMMA = Flavor and Extract Manufacturers' Association; K<sub>ow</sub> = octanol-water partition coefficient (log scale); K<sub>p</sub> = skin penetration coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated; P&G = Procter & Gamble Co.; RIFM = Research Institute for Fragrance Materials.

\*Mathematically estimated concentration of test chemical necessary to induce a threefold stimulation index (SI).

<sup>†</sup>Value estimated.

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Table 1. Continued

Chemical Structure	CAS No.	MW	$\log K_{ow}$	Vehicle	LLNA %	LLNA %	LLNA %	LLNA %	LLNA %	LLNA %	LLNA %	LLNA EC <sub>50</sub> * (%)	Potency Category	Nonsensitizer	P&G (unpublished)	Reference			
	55846-68-9	-0.97	5.35	336.47	Acetone	10.0	20.0	40.0	—	—	1.6	1.2	1.6	—	—	NC	P&G (unpublished)		
	167998-73-4	-1.84	3.14	221.32	Acetone	10.0	20.0	40.0	—	—	7.0	22.1	22.4	—	—	8.3 <sup>†</sup>	Moderate	P&G (unpublished)	
	137-26-8	-3.36	1.17	240.42	AOO	2.5	5.0	10.0	—	—	—	2.4	2.9	5.1	—	—	5.2	Moderate	Baskett DA et al. <sup>43</sup>
	108-77-0	-3.29	0.78	184.41	AOO	1.0	2.5	5.0	—	—	21.8	28.9	34.0	—	—	0.09 <sup>†</sup>	Extreme	Ashby J et al. <sup>21</sup>	

AOO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMF = dimethylformamide; DMSO = dimethyl sulfoxide; EtOH = ethanol; FEMA = Flavor and Extract Manufacturers' Association; K<sub>ow</sub> = octanol-water partition coefficient (log scale); K<sub>p</sub> = skin penetration coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated; P&G = Procter & Gamble Co.; RIEM = Research Institute for Fragrance Materials.

\*Mathematically estimated concentration of test chemical necessary to induce a threefold stimulation index (SI).  
†Value estimated.

Table I. Continued

Chemical Structure	CAS No.	Log K <sub>ow</sub>	Vehicle	LNNA EC <sub>50</sub> * (%)	Potency Category	Reference
4,4,4-Trifluoro-1-phenylbutane-1,3-dione	326-06-7	-2.27	2.52	219.18	Acetone	10.0 20.0 40.0
1-(3,4,5-Trimethoxyphenyl)-4-dimethylpentane-1,3-dione	135099-98-8	-2.78	2.47	297.37	Acetone	10.0 20.0 40.0
1,1,3-Trimethyl-2-formycyclohexa-2,4-diene (Safranal)	116-26-7	-1.83	2.54	150.22	AOO	0.5 1.0 2.5 5.0 10.0
3,5,5-Trimethylhexanoyl chloride	36727-29-4	-1.99	2.54	176.69	AOO	5.0 10.0 25.0
Undec-10-enal	112-45-8	-1.77	2.79	168.28	AOO	5.0 10.0 25.0 50.0 75.0

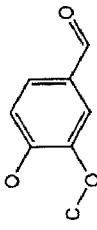
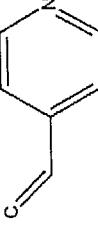
AOO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMF = dimethylformamide; DMSO = dimethyl sulfoxide; EtOH = ethanol; FEMA = Flavor and Extract Manufacturers' Association; K<sub>ow</sub> = octanol-water partition coefficient (log scale); K<sub>p</sub> = skin penetration coefficient (log scale); LNNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated; P&G = Procter & Gamble Co.; RIFM = Research Institute for Fragrance Materials.

\*Mathematically estimated concentration of test chemical necessary to induce a threefold stimulation index (SI).

<sup>†</sup>Value estimated.

(continued on next page)

Table 1. Continued

Chemical Structure	CAS No.	Log K <sub>ow</sub>	Log K <sub>p</sub>	MW	Vehilce	LLNA %	LLNA %	LLNA %	LLNA %	LLNA EC <sub>50</sub> * (%)	Potency Category	Reference
	121-33-5	-2.74	1.28	152.15	AOO	2.5	5.0	10.0	25.0	50.0	—	NC
	75-35-4	-2.28	1.45	96.94	AOO	10.0	25.0	50.0	—	—	0.8	0.9
	1337-81-1	-2.08	1.80	105.14	AOO	2.5	5.0	10.0	—	—	7.4	14.2
<i>t</i> Value estimated.												Ashby I et al. <sup>21</sup>

AOO = acetone and olive oil (4:1); CAS = Chemical Abstract Service number; DEP = diethyl phthalate; dH<sub>2</sub>O = distilled water; DMF = dimethylformamide; EtOH = ethanol; FEMA = Flavor and Extract Manufacturers' Association; K<sub>ow</sub> = octanol-water partition coefficient (log scale); K<sub>p</sub> = skin penetration coefficient (log scale); LLNA = local lymph node assay; MEK = methyl ethyl ketone; MW = molecular weight; NC = not calculated; P&G = Procter & Gamble Co.; RIFM = Research Institute for Fragrance Materials.

\*Mathematically estimated concentration of test chemical necessary to induce a threefold stimulation index (SI).

<sup>t</sup>Value estimated.

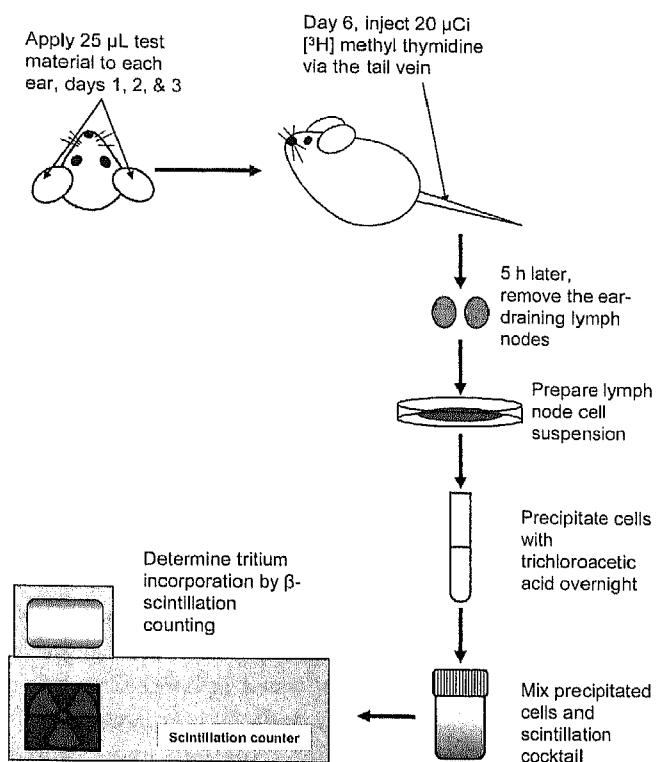


Figure 1. Schematic of local lymph node assay.

this article and in Table 1. The data are derived from previous studies that used LLNA methodology as described in the OECD guideline.<sup>7</sup> References for the sources of LLNA data for each of the chemicals are listed in Table 1.<sup>16,21–48</sup>

#### Potency Estimation in the Local Lymph Node Assay

The approach to the estimation of the relative skin sensitization potency of chemicals in the LLNA has been previously described in detail.<sup>10</sup> It is based on the mathematical estimation of the concentration of chemical necessary to obtain a threshold positive response (SI = 3); this is termed the EC<sub>3</sub> value. When the LLNA dose-response curve included concentrations that induced at least one SI greater than 3 and one SI less than 3, EC<sub>3</sub> values were calculated by linear interpolation. For chemicals that induced an SI greater than or equal to 3 at all concentrations tested, an EC<sub>3</sub> value was extrapolated from the two lowest doses used.<sup>49</sup> For conducting this extrapolation method, a dose response should be evident. The relative sensitizing potencies of the chemical allergens were categorized with a recently proposed arbitrary classification scheme.<sup>11</sup> The system, shown in Table 2, is composed of four sensitization potency categories based on EC<sub>3</sub> values. Compounds that did not induce a threefold increase at any concentration tested (generally > 20%) are categorized as nonsensitizing.

## Results

### Skin Sensitization Data Set: Chemical Information

Table 1 lists 211 chemical compounds along with their respective CAS numbers and two-dimensional chemical structures. It is clear from review of the structures themselves that the data set embraces the wide chemical diversity known to exist among skin allergens. For example, aldehydes, ketones, aromatic amines, quinones, and acrylates are represented in the data set. It is also known that the chemical data set lists compounds that demonstrate different protein reactivity mechanisms (eg, Schiff base, Michael addition). The physicochemical diversity of the allergens is reflected also by the log K<sub>O/W</sub> values, which range from less than -3 to greater than 5 (Table 3 and Fig 2). The majority of the allergens (92.3%), however, have log K<sub>O/W</sub> values in the ranges of -1 to 3 and 3 to 5. This is not surprising because chemicals with these ranges are known to be very good and good skin permeants, respectively.<sup>19,50–53</sup> The potency of an allergen does not seem to be associated with any one log K<sub>O/W</sub> range (eg, extreme allergens are represented in four of the five log K<sub>O/W</sub> categories). The nonallergens represented in the database have a log K<sub>O/W</sub> distribution similar to that of the allergens. Similar ranges of log K<sub>p</sub> values are evident for both the allergens and nonallergens (data not shown).

Not surprisingly, all of the allergens listed in Table 1 have an MW of less than 500 D, which is consistent with what has been published previously.<sup>54</sup> The MW distribution of the allergens shows that the majority are in the ranges of 100 to 200 D (52.7%) and 200 to 300 D (26.6%) (Table 4 and Fig 3). It is generally believed that chemical allergens have low MWs (< 500 D) and log K<sub>O/W</sub> values of greater than 1, which are thought to favor the penetration of the chemical across the lipid-rich stratum corneum.<sup>55</sup> For the most part, all potency categories are associated with the MW cutoff ranges depicted in Table 4. The distribution of the nonallergens in the database is similar to that of the allergens.

Table 2. Classification of Relative Skin Sensitization Potency by Local Lymph Node Assay EC<sub>3</sub> Values

EC <sub>3</sub> Value (%)	Potency Classification
≥ 10 to ≤ 100	Weak
≥ 1 to < 10	Moderate
≥ 0.1 to < 1	Strong
< 0.1	Extreme

**Table 3.** Log K<sub>O/W</sub> (Log P) Distribution of Allergens by Potency Classification

Log K <sub>O/W</sub> Range	Skin Penetration	Allergen Potency* Distribution
< -3	Very poor skin permeant, very hydrophobic	0
≥ -3 to -1	Poor skin permeant, hydrophilic	4 (2E, 1S, 0M, 1W)
≥ -1 to 3	Very good skin permeant, amphiphilic	113 (7E, 17S, 54M, 35W)
≥ 3 to 5	Good skin permeant, slightly lipophilic	43 (2E, 3S, 11M, 27W)
≥ 5	Moderate skin permeant but decreasing with increasing log K <sub>O/W</sub>	9 (2E, 0S, 4M, 3W)

\*Designated as E (extreme), S (strong), M (moderate), and W (weak).

### Skin Sensitization Data Set: Biologic Data

The LLNA data for each of the 211 chemicals are displayed in Table 1. The data set includes weak, moderate, strong, and extreme skin sensitizers, as well as nonsensitizers. The LLNA data in this database were from studies that used the standard LLNA protocol as described in OECD Test Guideline 429.<sup>7</sup> For some chemicals (such as those that formed the basis of the various interlaboratory trials), several different EC<sub>3</sub> values were available albeit within a narrow range. In each of these cases, the data shown in Table 1 derive from one representative experiment that we feel reflects accurately the results obtained with the chemical. For all of the studies represented in Table 1, a common vehicle as described in OECD Test Guideline 429 was used (eg, acetone and olive oil 4:1 [AOO], and dimethyl sulfoxide). In instances in which the chemical was tested in more than one vehicle, the data set for AOO was selected.

The specific reference for the source of the LLNA data for each chemical is indicated in Table 1. The data set includes 42 nonsensitizers, 66 weak sensitizers, 69 moderate sensitizers, 21 strong sensitizers, and 13 extreme sensitizers, for a total of 211 compounds (Fig 4). For the nonsensitizers, materials that did not give a positive response in the LLNA up to the highest dose tested were included (data for nonsensitizers were most often tested at concentrations of > 20%). For positive chemicals in the LLNA database, EC<sub>3</sub> values (calculated from linear interpolation) range from as low as 0.003% (for the sensitizer oxazolone) to as high as 89% (for aniline). In Table 1, we have also included EC<sub>3</sub> values estimated by the extrapolation method; these are indicated by a footnote marker. Of course, caution should be used in interpreting and using extrapolated EC<sub>3</sub> values that have been estimated from less-than-perfect dose-response data. However, it is our opinion that these data can still provide important benchmark information to investigators who are developing alternative skin sensitization testing methods.

### Discussion

One major focus area for scientists from academia and industry over the past decade has been to develop alternative nonanimal test methods for skin sensitization testing.

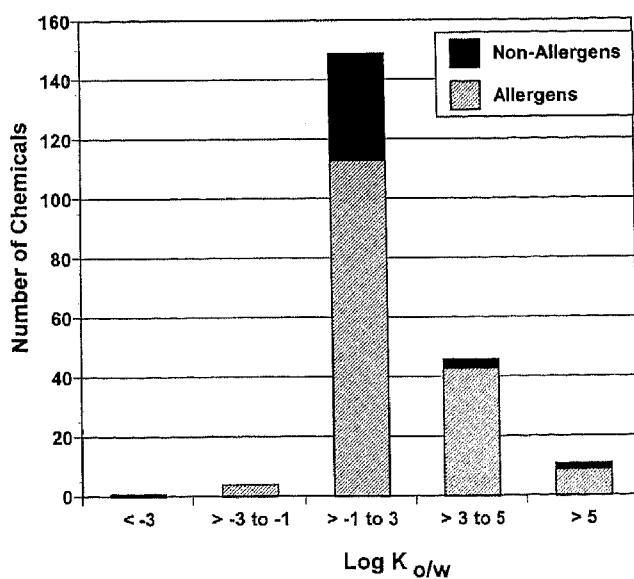


Figure 2. Distribution of log K<sub>O/W</sub> values of allergens and nonallergens.

**Table 4.** Molecular Weight Distribution of Allergen Potency

MW Range	Allergen Potency* Distribution
< 100	10 (0E, 2S, 3M, 5W)
≥ 100 to 200	89 (5E, 11S, 45M, 28W)
≥ 200 to 300	45 (7E, 4S, 12M, 22W)
≥ 300 to 400	23 (1E, 3S, 8M, 11W)
≥ 400	2 (0E, 1S, 1M, 0W)

MW = molecular weight.

\*Designated as E (extreme), S (strong), M (moderate), and W (weak).

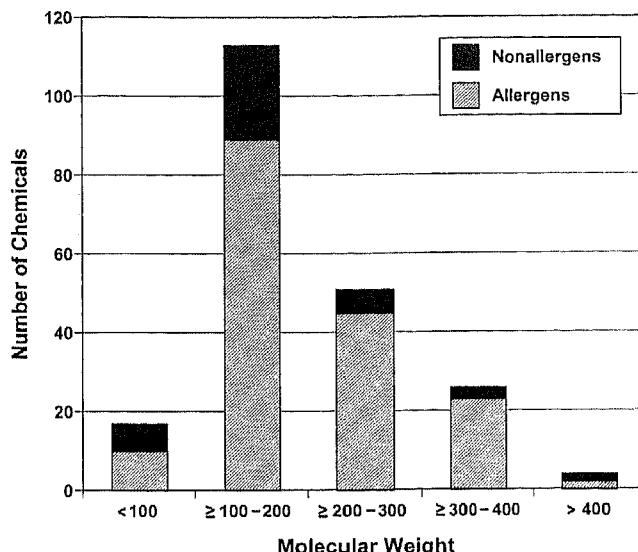


Figure 3. Distribution of molecular weights of allergens and non-allergens.

It is an enormous challenge to reproduce accurately *in vitro* the complex immunobiologic mechanisms that act in concert to permit skin sensitization. For example, numerous cell types (eg, T lymphocytes, Langerhans cells, keratinocytes) and a plethora of immune and inflammatory mediators (including cytokines and chemokines) are involved in the initiation and expression of an ACD response. The key to success will be the development of a test method or methods that incorporate all aspects of our understanding of the chemistry and biology of contact allergy.

Whatever new methods are developed, whether they are *in silico* quantitative structure activity system models or *in vitro* cell-based methods, they will be required to have their performance characteristics assessed, particularly their sensitivity, selectivity, and overall accuracy. This in turn requires the selection of a robust chemical data set to interrogate and calibrate the method.<sup>49,56</sup> One essential criterion is that the activity of each of the chemicals used must be supported by relevant and reliable *in vivo* data of high quality. In addition, it is important to select chemicals that display a wide range of potencies and that represent the relevant classes of chemicals and physical properties of the materials known to cause the specific endpoint. To this end, we compiled a data set of chemicals to be used for evaluating alternative approaches to skin sensitization testing that meet these criteria and that encompass the chemical and biologic diversity of chemicals known to cause skin sensitization in animals and/or in humans.

The chemicals listed in Table 1 have clearly different protein reaction mechanisms (eg, Schiff base formation,

Michael addition) and include representatives from various chemical classes of materials, including aldehydes, ketones, diketones, acrylates, and aromatic amines. The two-dimensional structure and CAS number of each compound is provided to aid investigators in obtaining the correct materials. It is well known that skin allergens must have a relatively low MW ( $\leq 500$  D)<sup>54</sup> and appropriate physicochemical properties (eg, lipophilicity).<sup>27,57</sup> The chemicals compiled in this data set demonstrate a distribution of  $\log K_{O/W}$  values and MWs that are consistent with chemicals known to be good to very good skin penetrants (see Tables 3 and 4 and Figs 2 and 3). It is known also that chemicals of equal molecular weight can have very different skin penetration potentials, depending on their lipophilicity. For example, most skin permeants are amphiphilic (or amphipathic) chemicals with a trend toward slight lipophilicity. It is also interesting that the allergenic potency of the chemicals compiled in the database are not associated with any one particular  $\log K_{O/W}$  or MW category (see Tables 3 and 4). The nonallergens in the database show similar distributions for the  $\log K_{O/W}$  and MW categories.

In addition to chemical diversity, it is particularly important that the enormous range of potencies known for skin allergens be encompassed. It is believed that differences between contact allergens with respect to their relative skin sensitizing potencies can span four or more orders of magnitude.<sup>11,15,58</sup> The chemicals compiled in this LLNA database display a range of potencies, which is referred to as the EC3 value.<sup>10</sup> The preferred method for deriving EC3 values is linear interpolation using the data points (concentration and SI) immediately above and below the SI value of 3 on the LLNA dose-response curve. In addition, we employed a method for estimating EC3 values that can be applied in instances in which none of the

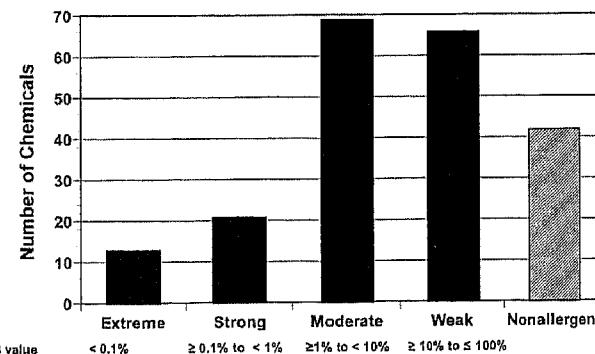


Figure 4. European Centre for Ecotoxicology and Toxicology of Chemicals (ECETOC) potency category distribution for allergens.

tested concentrations results in an SI of less than 3 and in which there is clear evidence of a dose response.<sup>49</sup> Although caution needs to be taken when employing this extrapolation method to obtain an EC3 value estimate, the method can provide information regarding likely threshold values and may obviate the need for repeat animal testing. It is also important to note that the data sets provided in this article are representative of the types of dose responses observed in the assay. For some chemicals (eg, camphorquinone), a decrease in SI values at the highest concentration tested may be indicative of local or systemic toxicity. Additional information such as body weight data can provide guidance in interpreting these types of responses.

The LLNA EC3 values listed in Table 1 show an enormous range of potency, covering four orders of magnitude, from weak allergens to extreme allergens. Each of the ECETOC potency categories established by the European Centre for Ecotoxicology and Toxicology of Chemicals<sup>11,12</sup> is well represented among the 211 chemicals in the data set (ie, 42 nonsensitizers, 66 weak sensitizers, 69 moderate sensitizers, 21 strong sensitizers, and 13 extreme sensitizers) (see Fig 4). Our experience is that the allergenic potency for mice as determined with the LLNA is similar to that known for humans.<sup>16,17,59</sup>

In terms of hazard identification (to the standards required globally for regulatory classification), the LLNA demonstrates the same accuracy as the Buehler and guinea pig maximization tests for predicting the human response.<sup>8</sup> Thus, there are a few compounds that demonstrate discordance between the LLNA results and human experience. For completeness, we chose to include those chemicals as well in this data set. For example, benzocaine, methyl 4-hydroxybenzoate (methylparaben), streptomycin sulfate, and vanillin are chemicals that are not detected as skin sensitizers in the LLNA but that have been reported as human allergens.<sup>16,17,25,28,35</sup> It is important to note that none of these chemicals that escaped detection in the LLNA are considered significant human contact allergens, nor do they typically give positive results in guinea pig tests. In addition, there are chemicals that give positive responses in the LLNA but are not considered skin allergens in humans. For example, isopropyl myristate and sodium lauryl sulfate yielded positive LLNA responses (albeit weak) whereas human experience suggests that these compounds are nonallergens.<sup>28,33</sup>

It is well known that some chemical allergens require biotransformation to be capable of initiating a skin sensitization response *in vivo*.<sup>27</sup> Thus, chemicals that are known to undergo activation or metabolism in the skin to acquire

reactivity are represented in this data set. (Such chemicals are called prohaptens; for example, eugenol and iso-eugenol are considered to be prohaptens.)<sup>27,42</sup> Because it is known that some chemical allergens must undergo biotransformation prior to reacting with proteins/peptides, it will be critical to incorporate a metabolism component into alternative assays in order to address these types of molecules.

An important measure of any new in vitro test method will be the extent of its usefulness in the skin sensitization risk assessment process. Although it would be of value to have in vitro methods available to assess the skin sensitization hazard of novel chemicals, it would be more valuable to have available methods that can extrapolate that hazard as risk to humans.<sup>1</sup> In addition to determining skin sensitization potential (hazard), the LLNA yields important information for assessing sensitization risk, the relative allergenic potency of a chemical.<sup>15,17,28</sup> Therefore, it will be useful to compare any new in vitro skin sensitization test method to the murine LLNA for its ability to determine allergenic potency. On the other hand, in vitro methods that can be used only for hazard identification would still be valuable because they can be used for screening purposes and will help to reduce the need for animal testing.

The list of chemicals contained in the data set represents both the chemical and biologic diversity known to exist for chemical allergens and nonallergens. It is our hope that publishing this LLNA data set will provide investigators with key information needed to accelerate the development and subsequent evaluation of alternative test methods and ultimately reduce the reliance on animals for assessing the skin sensitization potential of new chemicals.

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